

## WEST Search History

DATE: Sunday, November 02, 2003

Set Name   Query  
side by side

Hit Count   Set Name  
result set

*DB=USPT,PGPB,JPAB,EPAB,DWPI; THES=ASSIGNEE; PLUR=YES;  
OP=ADJ*

L7	L6 and cetane	5	L7
L6	L5 and cobalt	9	L6
L5	L3 and (carbon support\$3 or carbon near1 support\$3)	9	L5
L4	L3 and (carbon support or carbon near1 support)	8	L4
L3	L2 and (synthesis gas or carbon monoxide near1 hydrogen)	557	L3
L2	L1 and fischer near1 tropsch	744	L2
L1	diesel fuel or diesel distillate	16083	L1

END OF SEARCH HISTORY

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FILE COVERS 1907 - 2 Nov 2003 VOL 139 ISS 19  
FILE LAST UPDATED: 31 Oct 2003 (20031031/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> s diesel fuel or diesel distillate

37616 DIESEL  
408 DIESELS  
37664 DIESEL  
(DIESEL OR DIESELS)  
322317 FUEL  
150555 FUELS  
369966 FUEL  
(FUEL OR FUELS)  
16961 DIESEL FUEL  
(DIESEL(W) FUEL)  
37616 DIESEL  
408 DIESELS  
37664 DIESEL  
(DIESEL OR DIESELS)  
39703 DISTILLATE  
13870 DISTILLATES  
47825 DISTILLATE  
(DISTILLATE OR DISTILLATES)  
78 DIESEL DISTILLATE  
(DIESEL(W) DISTILLATE)

L1 16982 DIESEL FUEL OR DIESEL DISTILLATE

=> s l1 and Fischer tropisch

20971 FISCHER  
26 FISCHERS  
20982 FISCHER  
(FISCHER OR FISCHERS)  
6594 TROPSCH  
6478 FISCHER TROPSCH  
(FISCHER(W) TROPSCH)

L2 326 L1 AND FISCHER TROPSCH

=> s l2 and (synthesis gas or hydrogen (1a) carbon monoxide)

1089360 SYNTHESIS  
2 SYNTHESISES  
60253 SYNTHESSES  
1123625 SYNTHESIS  
(SYNTHESIS OR SYNTHESISES OR SYNTHESSES)  
1326106 GAS  
458873 GASES  
1491450 GAS  
(GAS OR GASES)

13871 SYNTHESIS GAS  
(SYNTHESIS(W) GAS)  
795652 HYDROGEN  
5232 HYDROGENS  
798590 HYDROGEN  
(HYDROGEN OR HYDROGENS)  
1023434 CARBON  
22805 CARBONS  
1031738 CARBON  
(CARBON OR CARBONS)  
154460 MONOXIDE  
943 MONOXIDES  
154958 MONOXIDE  
(MONOXIDE OR MONOXIDES)  
130447 CARBON MONOXIDE  
(CARBON(W) MONOXIDE)  
9050 HYDROGEN (1A) CARBON MONOXIDE  
L3 99 L2 AND (SYNTHESIS GAS OR HYDROGEN (1A) CARBON MONOXIDE)

=> s l3 and carbon (2a) support  
1023434 CARBON  
22805 CARBONS  
1031738 CARBON  
(CARBON OR CARBONS)  
370014 SUPPORT  
103439 SUPPORTS  
439579 SUPPORT  
(SUPPORT OR SUPPORTS)  
4078 CARBON (2A) SUPPORT  
L4 1 L3 AND CARBON (2A) SUPPORT

=> s l3 and activated carbon  
418475 ACTIVATED  
1023434 CARBON  
22805 CARBONS  
1031738 CARBON  
(CARBON OR CARBONS)  
37762 ACTIVATED CARBON  
(ACTIVATED(W) CARBON)  
L5 1 L3 AND ACTIVATED CARBON

=> s l4 or l5  
L6 2 L4 OR L5

=> s l6 and cobalt  
326350 COBALT  
93 COBALTS  
326354 COBALT  
(COBALT OR COBALTS)  
L7 2 L6 AND COBALT

=> d l7 ibib ab 1-2

L7 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2003 ACS on STN  
ACCESSION NUMBER: 2003:551193 CAPLUS  
DOCUMENT NUMBER: 139:103471  
TITLE: Process for direct synthesis of **diesel**  
**distillates** with high quality from  
**synthesis gas** through  
**Fischer-Tropsch** synthesis  
INVENTOR(S): Ding, Yunjie; Ma, Wenping; Lu, Yuan; Lin, Liwu  
PATENT ASSIGNEE(S): Peop. Rep. China  
SOURCE: U.S. Pat. Appl. Publ., 8 pp.  
CODEN: USXXCO

DOCUMENT TYPE: Patent  
LANGUAGE: English  
FAMILY ACC. NUM. COUNT: 1  
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2003134912	A1	20030717	US 2002-52485	20020117

PRIORITY APPLN. INFO.: US 2002-52485 20020117

AB **Diesel fuels** or blending stocks having excellent lubricity, oxidative stability and high cetane no. are directly produced from **synthesis gas** over **activated carbon** supported **cobalt** based **Fischer-Tropsch** catalyst under the condition of temp. within the range of 120.degree. to 400.degree., reaction pressure within the range of 0.5 to 10.0 MPa, vol. hourly space velocity of a mixt. of **hydrogen** and **carbon monoxide** within the range of 100 to 5000, the mole ratio of **hydrogen** to **carbon monoxide** within the range of 1 to 4. **Diesel fuels** contg. at least 95 wt.% paraffins with an iso to normal ratio of about 0.03 to 0.3, <50 ppm (wt.) of sulfur and nitrogen, less than about 2 wt.% unsaturates, and about 0.001 to less than 0.3 wt % oxygen were obtained by sepg. the **Fischer-Tropsch** product into a lighter (180.degree. to 245.degree. fraction) and heavier fractions (245.degree. to 380.degree. fraction) utilizing a rough flash, and combining the 180.degree. to 245.degree. portion of the lighter product with the 245.degree. to 380.degree. fraction in desired ratios.

L7 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2003 ACS on STN  
ACCESSION NUMBER: 1985:424785 CAPLUS  
DOCUMENT NUMBER: 103:24785  
TITLE: Conversion of syngas to liquid motor fuels  
INVENTOR(S): Rabo, Jule Anthony; Coughlin, Peter Kevin  
PATENT ASSIGNEE(S): Union Carbide Corp. , USA  
SOURCE: Eur. Pat. Appl., 54 pp.  
CODEN: EPXXDW  
DOCUMENT TYPE: Patent  
LANGUAGE: English  
FAMILY ACC. NUM. COUNT: 1  
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 140365	A1	19850508	EP 1984-113046	19841030
EP 140365	B1	19870819		
R: AT, BE, DE, FR, GB, IT, LU, NL, SE				
AT 29021	E	19870915	AT 1984-113046	19841030
AU 8434883	A1	19850509	AU 1984-34883	19841031
AU 577067	B2	19880915		
ZA 8408513	A	19850626	ZA 1984-8513	19841031
BR 8405576	A	19850910	BR 1984-5576	19841031
CA 1225982	A1	19870825	CA 1984-466715	19841031
US 4652538	A	19870324	US 1985-780259	19850926

PRIORITY APPLN. INFO.: US 1983-547668 19831101  
EP 1984-113046 19841030

AB A **Fischer-Tropsch** catalyst on a cocatalyst-support (hydrophobic zeolite) was used for the conversion of syngas [60:30:10 H-CO-Ar (added as a tracer)] to motor-fuel range C5+ hydrocarbons. Thus, a catalyst contg. .apprx.15% Co was prepd. by pptg. CoO on a steam-activated ultrahydrophobic (UHP) zeolite Y by stirring Co nitrate, Na2CO3, and zeolite suspension in water, filtering, drying, and redn. at 300.degree.. The catalyst was used to convert 50.87% syngas (300 psig, 251-253.degree., .apprx.400 h-1 space velocity) to C5+ hydrocarbons at 77.99% selectivity.



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NEWS 8 AUG 18 FROSTI and KOSMET enhanced with Simultaneous Left and Right  
Truncation  
NEWS 9 AUG 18 Simultaneous left and right truncation added to ANABSTR  
NEWS 10 SEP 22 DIPPR file reloaded  
NEWS 11 SEP 25 INPADOC: Legal Status data to be reloaded  
NEWS 12 SEP 29 DISSABS now available on STN  
NEWS 13 OCT 10 PCTFULL: Two new display fields added  
NEWS 14 OCT 21 BIOSIS file reloaded and enhanced  
NEWS 15 OCT 28 BIOSIS file segment of TOXCENTER reloaded and enhanced  
  
NEWS EXPRESS OCTOBER 01 CURRENT WINDOWS VERSION IS V6.01a, CURRENT  
MACINTOSH VERSION IS V6.0b(ENG) AND V6.0Jb(JP),  
AND CURRENT DISCOVER FILE IS DATED 23 SEPTEMBER 2003  
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NEWS WWW CAS World Wide Web Site (general information)

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NEWS	6	Feb 26	PCTFULL now contains images
NEWS	7	Mar 04	SDI PACKAGE for monthly delivery of multifile SDI results
NEWS	8	Mar 24	PATDPAFULL now available on STN
NEWS	9	Mar 24	Additional information for trade-named substances without structures available in REGISTRY
NEWS	10	Apr 11	Display formats in DGENE enhanced
NEWS	11	Apr 14	MEDLINE Reload
NEWS	12	Apr 17	Polymer searching in REGISTRY enhanced
NEWS	13	SEP 09	CA/CAPLUS records now contain indexing from 1907 to the present
NEWS	14	Apr 21	New current-awareness alert (SDI) frequency in WPIDS/WPINDEX/WPIX
NEWS	15	Apr 28	RDISCLOSURE now available on STN
NEWS	16	May 05	Pharmacokinetic information and systematic chemical names added to PHAR
NEWS	17	May 15	MEDLINE file segment of TOXCENTER reloaded
NEWS	18	May 15	Supporter information for ENCOMPPAT and ENCOMPLIT updated
NEWS	19	May 19	Simultaneous left and right truncation added to WSCA
NEWS	20	May 19	RAPRA enhanced with new search field, simultaneous left and right truncation
NEWS	21	Jun 06	Simultaneous left and right truncation added to CBNB
NEWS	22	Jun 06	PASCAL enhanced with additional data
NEWS	23	Jun 20	2003 edition of the FSTA Thesaurus is now available
NEWS	24	Jun 25	HSDB has been reloaded
NEWS	25	Jul 16	Data from 1960-1976 added to RDISCLOSURE
NEWS	26	Jul 21	Identification of STN records implemented
NEWS	27	Jul 21	Polymer class term count added to REGISTRY
NEWS	28	Jul 22	INPADOC: Basic index (/BI) enhanced; Simultaneous Left and Right Truncation available
NEWS	29	AUG 05	New pricing for EUROPATFULL and PCTFULL effective August 1, 2003
NEWS	30	AUG 13	Field Availability (/FA) field enhanced in BEILSTEIN
NEWS	31	AUG 15	PATDPAFULL: one FREE connect hour, per account, in September 2003
NEWS	32	AUG 15	PCTGEN: one FREE connect hour, per account, in September 2003
NEWS	33	AUG 15	RDISCLOSURE: one FREE connect hour, per account, in September 2003
NEWS	34	AUG 15	TEMA: one FREE connect hour, per account, in September 2003
NEWS	35	AUG 18	Data available for download as a PDF in RDISCLOSURE
NEWS	36	AUG 18	Simultaneous left and right truncation added to PASCAL
NEWS	37	AUG 18	FROSTI and KOSMET enhanced with Simultaneous Left and Right Truncation

NEWS 38 AUG 18 Simultaneous left and right truncation added to ANABSTR

NEWS EXPRESS April 4 CURRENT WINDOWS VERSION IS V6.01a, CURRENT  
MACINTOSH VERSION IS V6.0b(ENG) AND V6.0Jb(JP),  
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FULL ESTIMATED COST	1.26	1.26

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FILE COVERS 1907 - 14 Sep 2003 VOL 139 ISS 12

FILE LAST UPDATED: 12 Sep 2003 (20030912/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> s synthesis gas or (carbon monoxide (1a) hydrogen)

1082174 SYNTHESIS

3 SYNTHESISES

59967 SYNTHESSES

1116309 SYNTHESIS

(SYNTHESIS OR SYNTHESISES OR SYNTHESSES)

1318368 GAS

456711 GASES

1483043 GAS

(GAS OR GASES)



# WEST Search History

DATE: Sunday, September 07, 2003

<u>Set Name</u> side by side	<u>Query</u>	<u>Hit Count</u>	<u>Set Name</u> result set
<i>DB=USPT,PGPB,JPAB,EPAB,DWPI; THES=ASSIGNEE; PLUR=YES; OP=ADJ</i>			
L14	direct synthesis with diesel	3	L14
L13	L11 and one-step	1	L13
L12	L11 and primary product near3 diesel	1	L12
L11	L10 and cobalt	50	L11
L10	L9 and synthesis gas	56	L10
L9	diesel fuel and activated carbon	397	L9
L8	L7 and activated carbon	0	L8
L7	direct near1 diesel fuel	26	L7
L6	L5 not l4	1	L6
L5	L3 and activated carbon and cobalt	3	L5
L4	L3 and direct synthesis	2	L4
L3	diesel distillate	48	L3
L2	direct near1 synthesis near2 diesel fuel	0	L2
<i>DB=USPT,PGPB; THES=ASSIGNEE; PLUR=YES; OP=ADJ</i>			
L1	one near1 step with diesel fuel	2	L1

END OF SEARCH HISTORY

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FILE 'HOME' ENTERED AT 21:32:41 ON 07 SEP 2003

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	ENTRY	SESSION
FULL ESTIMATED COST	0.21	0.21

FILE 'CAPLUS' ENTERED AT 21:32:52 ON 07 SEP 2003

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FILE COVERS 1907 - 7 Sep 2003 VOL 139 ISS 11

FILE LAST UPDATED: 5 Sep 2003 (20030905/ED)

This file contains CAS Registry Numbers for easy and accurate  
 substance identification.

=> s direct synthesis (1) diesel fuel

515742 DIRECT  
 6046 DIRECTS  
 521050 DIRECT  
 (DIRECT OR DIRECTS)  
 1081199 SYNTHESIS  
 3 SYNTHESISES  
 59953 SYNTHESES  
 1115331 SYNTHESIS  
 (SYNTHESIS OR SYNTHESISES OR SYNTHESES)  
 4069 DIRECT SYNTHESIS

(DIRECT(W) SYNTHESIS)  
 37135 DIESEL  
 407 DIESELS  
 37185 DIESEL  
 (DIESEL OR DIESELS)  
 319774 FUEL  
 149994 FUELS  
 367284 FUEL  
 (FUEL OR FUELS)  
 16744 DIESEL FUEL  
 (DIESEL(W) FUEL)  
 L1 4 DIRECT SYNTHESIS (L) DIESEL FUEL

=> d 11 ibib ab 1-4

L1 ANSWER 1 OF 4 CAPLUS COPYRIGHT 2003 ACS on STN  
 ACCESSION NUMBER: 2003:551193 CAPLUS  
 DOCUMENT NUMBER: 139:103471  
 TITLE: Process for direct synthesis of diesel distillates  
 with high quality from synthesis gas through  
 Fischer-Tropsch synthesis  
 INVENTOR(S): Ding, Yunjie; Ma, Wenping; Lu, Yuan; Lin, Liwu  
 PATENT ASSIGNEE(S): Peop. Rep. China  
 SOURCE: U.S. Pat. Appl. Publ., 8 pp.  
 CODEN: USXXCO  
 DOCUMENT TYPE: Patent  
 LANGUAGE: English  
 FAMILY ACC. NUM. COUNT: 1  
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2003134912	A1	20030717	US 2002-52485	20020117
PRIORITY APPLN. INFO.:			US 2002-52485	20020117

AB Diesel fuels or blending stocks having excellent lubricity, oxidative stability and high cetane no. are directly produced from synthesis gas over activated carbon supported cobalt based Fischer-Tropsch catalyst under the condition of temp. within the range of 120.degree. to 400.degree., reaction pressure within the range of 0.5 to 10.0 MPa, vol. hourly space velocity of a mixt. of hydrogen and carbon monoxide within the range of 100 to 5000, the mole ratio of hydrogen to carbon monoxide within the range of 1 to 4. Diesel fuels contg. at least 95 wt.% paraffins with an iso to normal ratio of about 0.03 to 0.3, <50 ppm (wt.) of sulfur and nitrogen, less than about 2 wt.% unsaturates, and about 0.001 to less than 0.3 wt % oxygen were obtained by sepg. the Fischer-Tropsch product into a lighter (180.degree. to 245.degree. fraction) and heavier fractions (245.degree. to 380.degree. fraction) utilizing a rough flash, and combining the 180.degree. to 245.degree. portion of the lighter product with the 245.degree. to 380.degree. fraction in desired ratios.

L1 ANSWER 2 OF 4 CAPLUS COPYRIGHT 2003 ACS on STN  
 ACCESSION NUMBER: 2002:274070 CAPLUS  
 DOCUMENT NUMBER: 137:172057  
 TITLE: Development of dimethyl ether synthesis technology and its diesel engine test  
 AUTHOR(S): Ohno, Yotaro; Ogawa, Takashi; Shikada, Tsutomu; Hayashi, Hiromasa  
 CORPORATE SOURCE: NKK Corporation, Tokyo, 100-8202, Japan  
 SOURCE: Catalysts in Petroleum Refining and Petrochemicals, Proceedings [of the] Annual Saudi-Japanese Symposium, 10th, Dhahran, Saudi Arabia, Nov. 19-20, 2000 (2000),

Meeting Date 2000, 128-137. King Fahd University of  
Petroleum & Minerals: Dhahran, Saudi Arabia.

CODEN: 69CLNP

DOCUMENT TYPE: Conference

LANGUAGE: English

AB Di-Me Ether(DME) is a clean and economical alternative fuel with properties similar to **diesel fuel**, and which can be produced from various resources as natural gas, coal or biomass through synthesis gas. A combustion test of DME with a four cylinder DI diesel engine of 3600 cc and a running test using a light truck with the same type of engine were carried out. No soot is emitted at any conditions and Nox is reduced, while DME consumption is same by calorie as that of **diesel fuel**. An innovative process of **direct synthesis** of DME from synthesis gas has been developed. A newly developed catalyst in a slurry phase reactor gave a high conversion and high selectivity of DME. Pilot scale plant (5 tons/day) testing has successfully started in 1999 with the Japanese government support. A feasibility study of DME Fuel System, which includes prodn. of DME from natural gas and its transportation to Japan, indicates that DME is economically competitive to conventional fuels.

REFERENCE COUNT: 10 THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS  
RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L1 ANSWER 3 OF 4 CAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 2000:90340 CAPLUS

DOCUMENT NUMBER: 132:336791

TITLE: Direct dimethyl ether synthesis from hydrogen and carbon monoxide

AUTHOR(S): Ogawa, Takashi; Ono, Masami; Okuyama, Keiichi; Aoki, Seiji; Tomura, Keiji; Shikada, Tsutomu; Inoue, Norio; Ohno, Yotaro

CORPORATE SOURCE: Environmental Plant System Lab, Kawasaki Research Center, Japan

SOURCE: NKK Technical Review (1999), 81, 13-17

CODEN: NTERED; ISSN: 0915-0544

PUBLISHER: NKK Corp., Intellectual Property Dep.

DOCUMENT TYPE: Journal

LANGUAGE: English

AB DME (di-Me ether) is a clean and easy-to-use fuel that is similar to clean and environmentally-friendly LPG. DME is also known as a new clean **diesel fuel** that generates no soot. NKK developed a valuable new technol. for the **direct synthesis** of DME from carbon monoxide and hydrogen. This technol. converts coal, coal bed methane or natural gas to the nontoxic, gas-liq. fuel.

REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS  
RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L1 ANSWER 4 OF 4 CAPLUS COPYRIGHT 2003 ACS on STN

ACCESSION NUMBER: 1999:585036 CAPLUS

DOCUMENT NUMBER: 131:259679

TITLE: Increase of the cetane number of diesel fuel

AUTHOR(S): Kotowski, Wlodzimierz; Fechner, Wolfgang

CORPORATE SOURCE: Instytut Ciekkiej Syntezy Organicznej w Kedzierzynie-Kozlu, Politechnika Opolska Opole, Opole, Pol.

SOURCE: Gospodarka Paliwami i Energia (1999), 47(6), 8-10

CODEN: GPENAS; ISSN: 0017-2413

PUBLISHER: Wydawnictwo SIGMA-NOT

DOCUMENT TYPE: Journal

LANGUAGE: Polish

AB Di-Me ether boosts the cetane no. of **diesel fuel** from approx. 50 to 104 and decreases nitrogen oxide emissions by 22.4 vol.%. .

It can be produced by either **direct synthesis** from synthesis gas or by methanol dehydration.

RECEIVED 740	NO. _____ for following.	<input checked="" type="checkbox"/> 1.18 Fees ( Issue )
		<input type="checkbox"/> Other _____
		<input type="checkbox"/> Credit _____

# WEST Search History

DATE: Sunday, November 02, 2003

<u>Set Name</u> side by side	<u>Query</u>	<u>Hit Count</u>	<u>Set Name</u> result set
<i>DB=USPT,PGPB,JPAB,EPAB,DWPI; THES=ASSIGNEE; PLUR=YES; OP=ADJ</i>			
L61	l60 and activated carbon	2	L61
L60	L59 and (oxygenate or oxygen)	37	L60
L59	L58 and paraffins	41	L59
L58	L57 and sulfur and nitrogen	41	L58
L57	L56 and space velocity	44	L57
L56	L55 and cetane number	68	L56
L55	L3 and cobalt	463	L55
L54	one step synthesis near3 diesel fuel	0	L54
L53	L52 and (activated carbon or carbon near1 support)	0	L53
L52	6274029.pn.	2	L52
L51	L50 and (activated carbon or carbon near1 support)	0	L51
L50	5689031.pn.	2	L50
L49	L48 and (activated carbon or carbon near1 support)	0	L49
L48	5522983.pn.	2	L48
L47	L46 and (activated carbon or carbon near1 support)	0	L47
L46	5378348.pn.	2	L46
L45	L44 and (activated carbon or carbon near1 support)	0	L45
L44	5324335.pn.	2	L44
L43	l41 and (activated carbon or carbon near1 support)	0	L43
L42	l41 and diesel	1	L42
L41	4992406.pn.	2	L41
L40	L39 and (activated carbon or carbon near1 support)	0	L40
L39	4992159.pn.	2	L39
L38	L37 and (activated carbon or carbon near1 support)	0	L38
L37	4579986.pn.	2	L37
L36	L34 and carbon support	0	L36
L35	L34 and activated carbon	0	L35
L34	4542122.pn.	2	L34
L33	L32 and (diesel fuel or diesel distillate)	0	L33
L32	L31 and activated carbon	1	L32
L31	4478954.pn.	2	L31



L30	direct synthesis near3 diesel fuel	0	L30
L29	L28 and fischer tropsch	5	L29
L28	activated carbon support	371	L28
L27	L26 not l6	11	L27
L26	L25 and fischer tropsch	15	L26
L25	L24 and (diesel fuel or diesel distillate)	64	L25
L24	support with activated carbon	2491	L24
L23	L22 not l6	1	L23
L22	L21 and (disel fuel or diesel distillate)	5	L22
L21	fischer tropsch and activated carbon	262	L21
L20	L19 and fischer	5	L20
L19	carbon supported cobalt	12	L19
L18	L17 and support	2	L18
L17	4617320.pn.	2	L17
L16	l13 not l6	15	L16
L15	direct synthesis near3 diesel distillate	2	L15
L14	direct synthesis near3 diesel fuel	0	L14
L13	L12 and space velocity	18	L13
L12	L11 and (oxygen or oxygenate)	48	L12
L11	L10 and sulfur and nitrogen	52	L11
L10	L9 and cetane	58	L10
L9	L8 and cobalt	226	L9
L8	l3 and support\$3 with carbon	236	L8
L7	L6 and cetane	5	L7
L6	L5 and cobalt	9	L6
L5	L3 and (carbon support\$3 or carbon near1 support\$3)	9	L5
L4	L3 and (carbon support or carbon near1 support)	8	L4
L3	L2 and (synthesis gas or carbon monoxide near1 hydrogen)	557	L3
L2	L1 and fischer near1 tropsch	744	L2
L1	diesel fuel or diesel distillate	16083	L1

END OF SEARCH HISTORY

**WEST**

Generate Collection

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L27: Entry 8 of 11

File: USPT

Jul 17, 1984

DOCUMENT-IDENTIFIER: US 4460710 A

TITLE: Catalytic method for synthesizing hydrocarbons

Brief Summary Text (2):

This invention relates to the synthesis of hydrocarbons by the reaction of carbon monoxide and hydrogen in the presence of a catalyst, commonly known as the Fischer-Tropsch synthesis. More particularly, this invention relates to novel catalysts for use in such process, methods for preparation of such catalysts, and methods for use of such catalysts.

Brief Summary Text (4):

The so-called Fischer-Tropsch synthesis wherein liquid aliphatic hydrocarbons, alcohols and minor amounts of aldehydes, fatty acids and ketones are produced by the hydrogenation of carbon monoxide has been known for about 60 years. Initially, alkalized iron turnings were utilized as the catalytic material. Typical effective catalysts are supported cobalt-thoria or supported iron catalysts. The reaction temperature is about 250.degree.-300.degree. C. and pressures range from 1 atm. to about 20 atm. A large commercial plant using iron catalysts is in operation in South Africa. Additionally, various methods for conducting the specific contacting of the reactants with one another and the catalytic material have been utilized, e.g. fixed bed, fluidized bed, etc. A thorough discussion of the chemistry of this immensely important reaction is set forth in "The Fischer-Tropsch and Related Syntheses" by Henry H. Storch, Norma Golumbic, Robert B. Anderson, published by John Wiley & Sons, New York, 1951.

Brief Summary Text (5):

Numerous attempts have been made to refine this synthesis in terms of improved effectiveness of the catalyst, product yield, improved production of more desirable product fractions, control of the product distribution, etc. Additionally, efforts have been made to achieve more stable catalysts. As a general rule, the materials which have been known to be effective as Fischer-Tropsch catalysts are extremely sensitive to air and moisture and consequently, must be used either shortly after preparation or prepared in situ.

Brief Summary Text (8):

We have discovered a novel catalytic material which can be used for the synthesis of hydrocarbons from carbon monoxide and hydrogen. This catalytic material is unique in both its physiochemical constitution as well as the properties which it exhibits. Thus, the catalytic composition of the present invention exhibits superior activity as compared to conventional Fischer-Tropsch type catalysts. In addition, such activity can be obtained in dilute slurry form which substantially improves the heat transfer factors involved in the Fischer-Tropsch synthesis. Furthermore, the catalytic composition of the present invention exhibits superior stability and can be stored for long periods of time in either a dry or slurry form. Finally, the catalytic composition of the present invention produces a very desirable product composed of a fraction of linear hydrocarbons ranging from C.sub.1 to C.sub.40 with a low degree of branching.

Brief Summary Text (10):

The method and use of the present invention is also unique as compared with conventional Fischer-Tropsch catalysts. Of great importance is the fact that this catalyst can be used in dilute slurry form and with dilute concentrations of gaseous

reactants to obtain high yields of desirable product fractions. This avoids the heat transfer problems commonly encountered with alternative contacting systems.

Brief Summary Text (11):

More particularly, the catalyst composition of the present invention is composed of palladium or platinum and cobalt supported on a solid phase. The solid phase material, commonly referred to as a support or carrier, may be chosen from any appropriate material such as: talc; dolomite; limestone; clay; activated carbon; zeolite; pumice; the oxides, hydroxides or carbonates of aluminum, silicon, zinc, chromium, magnesium, calcium, titanium, or zirconium; alumina; silica gel; kieselguhr; barium sulfate; or any inert material. This catalyst produces improved yields significantly greater than conventionally known systems. Additionally, the catalyst is able to operate effectively under wider ranges of pressure and temperature than the previously known catalysts and can also operate effectively under dilute feed gas conditions, that is in the presence of synthesis gas diluents or impurities such as nitrogen, so long as the ranges of carbon monoxide and hydrogen are within the ranges set forth below.

Drawing Description Text (8):

FIG. 7 is a gas chromatographic analysis of diesel fuel.

Detailed Description Text (24):

Of particular importance is the fact that despite the wide range of pressures and temperatures which can be used as well as the diluteness of the gaseous feed streams, the product obtained is composed of highly desirable fractions of hydrocarbons. Typically, for example, the product obtained with the present invention would have an analysis as shown by gas phase chromatography exemplified in FIG. 6. The distribution for diesel fuel is shown in FIG. 7. In particular, analysis of typical reaction products indicates a broad distribution of C.sub.1 to C.sub.40 paraffins. Only small amounts of C.sub.1 -C.sub.5 alcohols have been detected with the catalyst of the invention. For example, a reaction employing 3 g of the present catalyst in 100 ml of xylene carried out under three separate synthesis gas chargings yielded a light yellow solution and water. The xylene was distilled under vacuum, a yellow oil resulted. Infrared analysis of the reaction solution and oil indicated only a small amount of oxygenated product and olefins and no metal carbonyl. Integrated nuclear magnetic resonance spectra of the yellow oil, shown in FIG. 8, indicated highly linear paraffinic products of an average chain length of 18 with little or no aromatics, unsaturates, oxygenates or branched products. Gas phase analysis of the gaseous components from the cooled reactor, as determined by gas-solid chromatography employing a thermal conductivity detector, indicated methane (generally less than 10 weight percent of hydrocarbon in the product), ethane, propane, butane, and only small amounts of unsaturated hydrocarbons.

Detailed Description Text (25):

An additional experiment was carried out to determine the efficacy of the catalyst in accordance with the present invention in dilute synthesis gas feed conditions. In particular, a catalyst was prepared from 3.4 grams of dicobalt octacarbonyl and 1.0 grams 5% palladium on alumina in cyclohexane under 400 psi of nitrogen and 1200 psi of hydrogen. The reactor was heated to about 180.degree. C. However, no pressure drop was observed. On cooling the reactor, ammonia could not be detected in either the gas or liquid phases. Upon venting the gases and recharging with 500 psi of nitrogen and 500 psi of synthesis gas (2 H.sub.2 :1 CO) and thereafter heating, the catalyst exhibited normal activity for hydrocarbon synthesis. This is to be contrasted with ordinary Fischer-Tropsch catalysts which show a marked decrease in activity in the presence of a diluent.

Detailed Description Text (26):

That the catalyst of the present invention is substantially more reactive than conventionally known Fischer-Tropsch catalysts is shown in Table 1.

Detailed Description Text (27):

Table 1 shows rate comparisons of a number of conventional Fischer-Tropsch catalysts with catalysts of the present invention under isothermal conditions. The activity of the present catalysts as expressed in conversions per catalytic volume per unit time or of conversions per mole of metal atoms per unit time are superior by about two orders of magnitude as compared to those catalysts studied by the Bureau of Mines.

Detailed Description Text (28):

Of particular interest is the fact that the conventional catalysts for hydrocarbon synthesis are generally used at about atmospheric pressure. In contrast, the catalysts of the present invention have their best activity at pressures between about 300 to 500 psi. An experiment utilizing a conventional Fischer-Tropsch catalyst, such as, 100 Co : 18 ThO.sub.2 : 100 kieselguhr, under comparable conditions to those used with the present catalyst (cyclohexane slurry, 225.degree. C., 1200 psi cold synthesis gas pressure), show that the rate of gas consumption was more than 10 times faster with the present catalyst, while the product of the present catalyst contained less of the undesired oxygenates.

Detailed Description Paragraph Table (1):

TABLE 1		REPRESENTATIVE HYDROCARBON PRODUCTION	
RATES OF FISCHER TROPSCH CATALYSTS	Activity g prod. per Temperature Kg metal Catalyst		
.degree.C. per hour			Catalyst of Example 4
225.degree. 3000	Catalyst of Example 9	225.degree. 1080	Catalyst of the Present
125.degree. 40	Invention.sup.a Lurgi catalyst (10Fe:10Cu: 225.degree. 24 2K.sub.2		
CO.sub.3 :9Al.sub.2 O.sub.3 :30SiO.sub.2).	sup.b Brabag catalyst (100Fe:20Cu:		
225.degree. 10 20Zn:1K.sub.2 CO.sub.3).	sup.b Bureau of Mines 2A catalyst	195.degree.	
50 (100Co:18ThO.sub.2 :100 kieselguhr).	sup.c Pichler acid-promoted Ru/Al.sub.2 O.sub.3		
120.degree. 120 catalyst for polymethylene.	sup.d Kolbel slurry catalyst	268.degree.	
450 (100Fe:0.1Cu:0.05K.sub.2 O).	sup.e Vannice (5% Fe on glassy 235.degree. 4		
carbon).sup.f			sup.a 2.2 g catalyst, containing
1.2 g metal on low surface area (80-100 mesh) Al.sub.2 O.sub.3, 100 mL cyclohexane,			
1200 psi charge 2.1 syngas, 300 mL AE reactor, catalyst prepared in situ.	sup.b H. H.		
Storch, N. Golumbic, and R. B. Anderson, The FischerTropsch and Related Syntheses, p.			
308 (Table 86), Wiley, New York, 1951.	sup.c Ibid., p. 132 (Table 5).	sup.d H.	
Pichler and F. Bellstedt, Erdol u. Kohle 26, 560 (1973).	sup.e H. Kolbel, P.		
Ackermann, and F. Engelhardt, Erdol u. Kohle 9, 153, 225, 303 (1956).	sup.f M. A.		
Vannice, paper presented at 181st Am. Chem. Soc. Meet., Atlanta, GA, March 29-April			
3, 1981.			

# WEST Search History

DATE: Sunday, November 02, 2003

<u>Set Name</u> side by side	<u>Query</u>	<u>Hit Count</u>	<u>Set Name</u> result set
<i>DB=USPT,PGPB,JPAB,EPAB,DWPI; THES=ASSIGNEE; PLUR=YES; OP=ADJ</i>			
L27	L26 not l6	11	L27
L26	L25 and fischer tropsch	15	L26
L25	L24 and (diesel fuel or diesel distillate)	64	L25
L24	support with activated carbon	2491	L24
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L22	L21 and (disel fuel or diesel distillate)	5	L22
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L13	L12 and space velocity	18	L13
L12	L11 and (oxygen or oxygenate)	48	L12
L11	L10 and sulfur and nitrogen	52	L11
L10	L9 and cetane	58	L10
L9	L8 and cobalt	226	L9
L8	l3 and support\$3 with carbon	236	L8
L7	L6 and cetane	5	L7
L6	L5 and cobalt	9	L6
L5	L3 and (carbon support\$3 or carbon near1 support\$3)	9	L5
L4	L3 and (carbon support or carbon near1 support)	8	L4
L3	L2 and (synthesis gas or carbon monoxide near1 hydrogen)	557	L3
L2	L1 and fischer near1 tropsch	744	L2
L1	diesel fuel or diesel distillate	16083	L1

END OF SEARCH HISTORY

**WEST**

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**Search Results - Record(s) 1 through 5 of 5 returned.**☐ 1. Document ID: US 20030134912 A1

L7: Entry 1 of 5

File: PGPB

Jul 17, 2003

PGPUB-DOCUMENT-NUMBER: 20030134912

PGPUB-FILING-TYPE: new

DOCUMENT-IDENTIFIER: US 20030134912 A1

TITLE: Process for direct synthesis of diesel distillates with high quality from synthesis gas through fischer-tropsch synthesis

PUBLICATION-DATE: July 17, 2003

## INVENTOR-INFORMATION:

NAME	CITY	STATE	COUNTRY	RULE-47
Ding, Yunjie	Liaoning		CN	
Ma, Wenping	Liaoning		CN	
Lu, Yuan	Liaoning		CN	
Lin, Liwu	Liaoning		CN	

APPL-NO: 10/ 052485 [PALM]

DATE FILED: January 17, 2002

INT-CL: [07] B01 J 8/04, C07 C 27/06

US-CL-PUBLISHED: 518/715; 422/188

US-CL-CURRENT: 518/715; 422/188

REPRESENTATIVE-FIGURES: 1

## ABSTRACT:

Diesel fuels or blending stocks having excellent lubricity, oxidative stability and high cetane number are directly produced from synthesis gas over activated carbon supported cobalt based Fischer-Tropsch catalyst under the condition of temperature within the range of 120 to 400.degree. C., reaction pressure within the range of 0.5 to 10.0 MPa, volume hourly space velocity of a mixture of hydrogen and carbon monoxide within the range of 100 to 5000, the mole ratio of hydrogen to carbon monoxide within the range of 1 to 4. Diesel fuels containing at least 95 wt % paraffins with an iso to normal ratio of about 0.03 to 0.3, <50 ppm (wt) of sulfur and nitrogen, less than about 2 wt % unsaturates, and about 0.001 to less than 0.3 wt % oxygen were obtained by separating the Fischer-Tropsch product into a lighter (180 to 245 .degree. C. fraction) and heavier fractions (245 to 380.degree. C. fraction) utilizing a rough flash, and combining the 180 to 245.degree. C. portion of the lighter product with the 245 to 380.degree. C. fraction in desired ratios.

Full	Title	Citation	Front	Review	Classification	Date	Reference	Sequences	Attachments	Claims	KMC	Draw Desc
Image												

☐ 2. Document ID: US 20030121826 A1

L7: Entry 2 of 5

File: PGPB

Jul 3, 2003

PGPUB-DOCUMENT-NUMBER: 20030121826  
PGPUB-FILING-TYPE: new  
DOCUMENT-IDENTIFIER: US 20030121826 A1

TITLE: Activated carbon supported cobalt based catalyst for direct conversion of  
synthesis gas to diesel fuels

PUBLICATION-DATE: July 3, 2003

## INVENTOR-INFORMATION:

NAME	CITY	STATE	COUNTRY	RULE-47
Ding, Yunjie	Liaoning		CN	
Ma, Wenping	Liaoning		CN	
Lin, Liwu	Liaoning		CN	

APPL-NO: 10/ 034206 [PALM]  
DATE FILED: January 3, 2002

INT-CL: [07] C07 C 27/06, C10 G 11/04, B01 J 21/18

US-CL-PUBLISHED: 208/120.1; 518/714, 502/180  
US-CL-CURRENT: 208/120.1; 502/180, 518/714

## ABSTRACT:

Diesel fuels or blending stocks having high cetane number are produced from non-shifting Fischer-Tropsch processes, which directly convert carbon monoxide and hydrogen to diesel distillates over activated carbon supported cobalt based Fischer-Tropsch catalysts. The activated carbon supported cobalt based catalysts comprise a substantially high dispersion of at least one of a zirconium component, an cerium component, a ruthenium component or a potassium component in porous carbon and elemental cobalt either deposited thereon or substantially uniformly dispersed therein, wherein the concentration of activated carbon in the catalyst is from about 20 to about 90 percent by weight, based on the weight of the catalyst, the concentration of elemental cobalt in the catalyst is from about 4 to about 50 percent by weight, based on the weight of the catalyst, the total concentration of the zirconium component, the cerium component, or a combination thereof in the catalyst is from about 0.01 to about 20 percent by weight, based on the weight of the catalyst and calculated as the elemental metal or metals, and the total concentration of the ruthenium component, the potassium component, or a combination thereof in the catalyst is from about 0.01 to about 5.0 percent by weight, based on the weight of the catalyst and calculated as the elemental metal or metals. Activated carbon carrier has a surface area in the range of about 200-2000 m.sup.2/g, preferably 800-1500 m.sup.2/g, and a pore volume of 0.3 to 2.0 ml/g, preferably 0.35 to 0.75 ml/g, a distribution of pore diameter of 4 to 1000 .ANG., preferably 5 to 500 .ANG..

Full	Title	Citation	Front	Review	Classification	Date	Reference	Sequences	Attachments	Claims	KMIC	Draw Desc
Image												

☐ 3. Document ID: US 6566568 B1

L7: Entry 3 of 5

File: USPT

May 20, 2003

US-PAT-NO: 6566568

DOCUMENT-IDENTIFIER: US 6566568 B1

TITLE: Molecular averaging of light and heavy hydrocarbons

DATE-ISSUED: May 20, 2003

## INVENTOR-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY
Chen; Cong-Yan	Kensington	CA		

## ASSIGNEE-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY	TYPE CODE
Chevron U.S.A. Inc.	San Ramon	CA			02

APPL-NO: 10/ 027838 [PALM]

DATE FILED: December 19, 2001

INT-CL: [07] C07 C 6/08

US-CL-ISSUED: 585/310; 585/644, 585/646, 585/647

US-CL-CURRENT: 585/310; 585/644, 585/646, 585/647

FIELD-OF-SEARCH: 585/310, 585/644, 585/646, 585/647

PRIOR-ART-DISCLOSED:

## U.S. PATENT DOCUMENTS

PAT-NO	ISSUE-DATE	PATENTEE-NAME	US-CL
<u>3657109</u>	April 1972	Beyaert	
<u>3699035</u>	October 1972	Hughes et al.	
<u>3718576</u>	February 1973	Hughes et al.	
<u>3728410</u>	April 1973	Hughes	
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## FOREIGN PATENT DOCUMENTS

FOREIGN-PAT-NO	PUBN-DATE	COUNTRY	US-CL
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Thomas R. Hughes et al., Molecular Redistribution of Alkanes and Alkylbenzenes, Paper No. 87, pp. 87-1217 through 1228.  
Robert L. Burnett et al., Mechanism and Poisoning of the Molecular Redistribution Reaction of Alkanes with a Dual-Functional Catalyst System, Journal of Catalysis, Mar.



28, 1973, pp. 55-64, vol. 31, 1973 Academic Press, Inc.

ART-UNIT: 1764

PRIMARY-EXAMINER: Dang; Thuan D.

ATTY-AGENT-FIRM: Ellinwood; Steven R.

ABSTRACT:

A process for preparing a paraffinic product stream in the gasoline, middle distillate fuel and lube ranges from a C.sub.2-5 -containing feedstock and a C.sub.20 + paraffinic feedstock is described. The combined feedstocks are subjected to molecular averaging via dehydrogenation to form olefins, metathesis of the olefins, and rehydrogenation of the olefins to form paraffins. The product stream includes a fraction rich in paraffins the molecular weights of which are between those of the light and heavy paraffin feedstocks, plus some unconverted feeds. The product of the molecular averaging reaction can optionally be isomerized to improve the octane value, in the case of gasoline, or pour point, in the case of middle distillate fuels and lubes. The unconverted feedstocks can be recycled to extinction.

20 Claims, 3 Drawing figures

Full	Title	Citation	Front	Review	Classification	Date	Reference	Sequences	Attachments	KWIC	Draw Desc
Image											

☐ 4. Document ID: US 6369286 B1

L7: Entry 4 of 5

File: USPT

Apr 9, 2002

US-PAT-NO: 6369286

DOCUMENT-IDENTIFIER: US 6369286 B1

TITLE: Conversion of syngas from Fischer-Tropsch products via olefin metathesis

DATE-ISSUED: April 9, 2002

INVENTOR-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY
O'Rear; Dennis J.	Petaluma	CA		

ASSIGNEE-INFORMATION:

NAME	CITY	STATE	ZIP CODE	COUNTRY	TYPE CODE
Chevron U.S.A. Inc.	San Ramon	CA			02

APPL-NO: 09/ 517370 [PALM]

DATE FILED: March 2, 2000

INT-CL: [07] C07 C 6/00

US-CL-ISSUED: 585/644; 585/643, 585/324

US-CL-CURRENT: 585/644; 585/324, 585/643

FIELD-OF-SEARCH: 585/643, 585/644, 585/324, 585/254

PRIOR-ART-DISCLOSED:

U.S. PATENT DOCUMENTS

PAT-NO	ISSUE-DATE	PATENTEE-NAME	US-CL
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<u>2662090</u>	December 1953	Scharmann	
<u>2686195</u>	August 1954	McAdams et al.	
<u>2735862</u>	February 1956	Buchmann et al.	
<u>2850515</u>	September 1958	Riblett et al.	
<u>3634538</u>	January 1972	Steffgen	585/644
<u>3668268</u>	June 1972	Mulaskey	
<u>3723562</u>	March 1973	Heckelsberg	585/324
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2 050 859

PUBN-DATE

January 1981

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ART-UNIT: 1764

PRIMARY-EXAMINER: Griffin; Walter D.

ATTY-AGENT-FIRM: Burns, Doane, Swecker &amp; Mathis, LLP.

## ABSTRACT:

A process for preparing distillate fuel compositions from a C.sub.2-6 olefinic

fraction and a C.sub.20 + fraction via molecular averaging is described. The fractions can be obtained, for example, from Fischer-Tropsch reactions, and/or obtained from the distillation or other processing of crude oil. Molecular averaging converts the fractions to a product that includes a significant fraction in the C.sub.5-20 range that can be used for preparing a distillate fuel composition. The product is preferably isomerized to increase the octane value and lower the pour, cloud and smoke point. The product can also be hydrotreated and/or blended with suitable additives for use as a distillate fuel composition.

24 Claims, 1 Drawing figures

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TITLE: Converting synthesis gas to hydrocarbons with high diesel distillates content through Fischer-Tropsch process used as, e.g. diesel fuel heavier than gasoline, involves employing activated carbon supported cobalt-based catalyst

INVENTOR: DING, Y; LIN, L ; LU, Y ; MA, W

PRIORITY-DATA: 2002US-0052485 (January 17, 2002)

PATENT-FAMILY:

PUB-NO	PUB-DATE	LANGUAGE	PAGES	MAIN-IPC
US 20030134912 A1	July 17, 2003		008	B01J008/04

INT-CL (IPC): B01 J 8/04; C07 C 27/06

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